

EPA Method 8270 with Nitrogen Carrier Gas

Paul Macek

Shimadzu Scientific Instruments, Inc.

Southeast Regional Office

Durham, NC 27703

08 August, 2019

Method 8270 with Nitrogen Carrier Gas

A work in progress

This is progress Report Number 1

Method 8270 with Nitrogen Carrier Gas

- At this point we are concentrating on 2 items:
 - Chromatographic separation
 - Detection limits
- If we are not able to generate acceptable results in either area, there is no reason to continue the study

Why Nitrogen?

- **The helium shortage is real**
 - Rationing is already in place
 - High per tank cost
 - As high as \$1000 per tank
- **Hydrogen does not work for many 8270 targets**
 - Works well for some neutrals
 - Not so well for acids, bases, other neutrals
 - In-source reactions (e. g. nitrobenzene)
 - High background from contamination



Results with Hydrogen Carrier

8270 Compounds that work well with H₂ Carrier

1,3-Dichlorobenzene	Fluorene
1,4-Dichlorobenzene	N-Nitrosodiphenylamine
1,2-Dichlorobenzene	Hexachlorobenzene
Bis(2-Chloroisopropyl) ether	Phenanthrene
N-Nitrosodi-n-propylamine	Anthracene
1,2,4-Trichlorobenzene	Fluoranthene
Naphthalene	Pyrene
2-Methylnaphthalene	Benzo[a]anthracene
4-Chloroaniline	Chrysene
2-Chloronaphthalene	Benzo[b]fluoranthene
Acenaphthene	Benzo[k]fluoranthene
Dibenzofuran	Benzo[a]pyrene
4-Chlorophenyl phenyl ether	

Results with Hydrogen Carrier

8270 compounds produce poor results with H₂ carrier

N-Nitrosodimethylamine

Phenol

2-Chlorophenol

Benzyl alcohol

3&4-Methylphenol

2-Methylphenol

Hexachloroethane

2,4-Dimethylphenol

Bis(2-chloroethoxy)methane

2,4-Dichlorophenol

Isophorone

Hexachlorobutadiene

4-Chloro-3-methylphenol

Hexachlorocyclopentadiene

2,4,6-Trichlorophenol,

2,4,5-Trichlorophenol,

Indeno[1,2,3-cd]pyrene

Dibenzo(a,h)anthracene

Pentachlorophenol

Carbazole

Results with Hydrogen Carrier

8270 compounds produce very poor results with H₂ carrier

2-Nitrophenol

2-Nitroaniline

Dimethyl phthalate

2,6-Dinitrotoluene

3-Nitroaniline

2,4-Dinitrophenol

4-Nitrophenol

2,4-Dinitrotoluene

Diethyl phthalate

4-Nitroaniline

2-Methyl-4,6-dinitrophenol

Di-n-butyl phthalate

Butylbenzyl phthalate

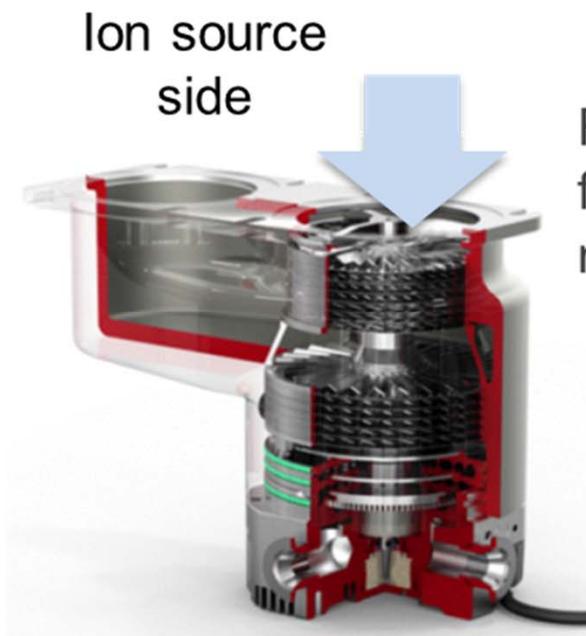
3,3'-Dichlorobenzidine

Bis(2-ethylhexyl) phthalate

Di-n-octyl phthalate

Can a GC/MS Pump Nitrogen?

- **Most cannot – Especially older units**
- **Newest instruments can if equipped with the latest pump**
 - Differential pumping helps too
- **Shimadzu GCMS QP-2020NX is equipped to pump nitrogen**
 - **Edwards nEXT-200/200D**
 - Differential pumping



Further noise reduction for H₂ and N₂ with the new TMP

**Dual inlet differential evacuation
+
High efficient TMP**

Shimadzu GCMS QP-2020



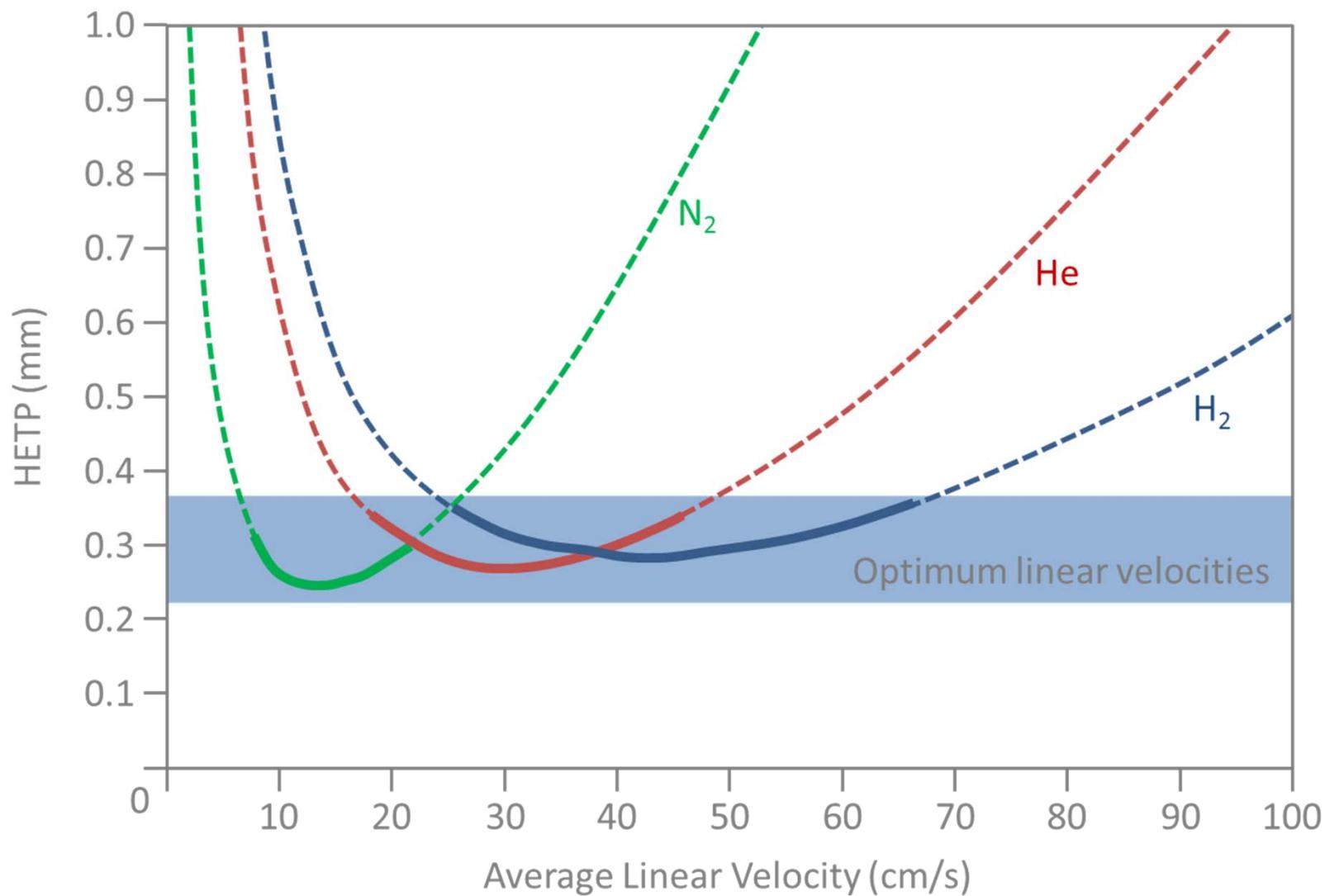
What negative effects did we expect to see?

- **Longer Chromatographic Runs**
 - Van Deemter Plot
- **Reduced sensitivity**
 - Caused by higher source pressure
 - 7X reduction in sensitivity expected
- **Band broadening on lighter compounds caused by low flow through the injection port**

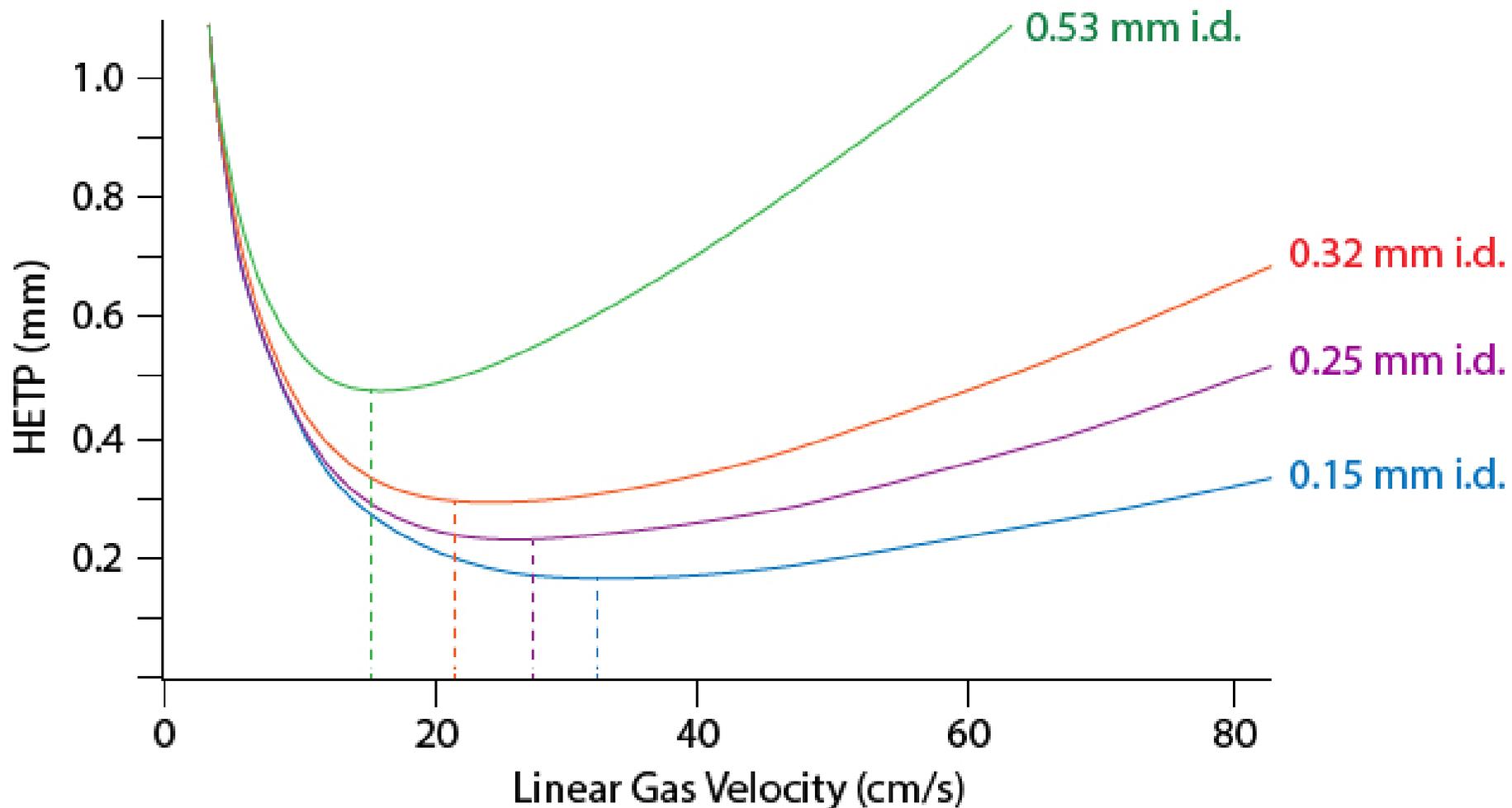
What positive effects did we expect to see?

- **May be able to use smaller ID columns**
 - Optimum linear velocity increases as column ID decreases
 - Lower flow minimizes effect of nitrogen on sensitivity
- **Less chromatographic impact on active compounds than H₂ carrier**
- **Less impact from contamination than with H₂ carrier**
- **No in-source reactions expected**

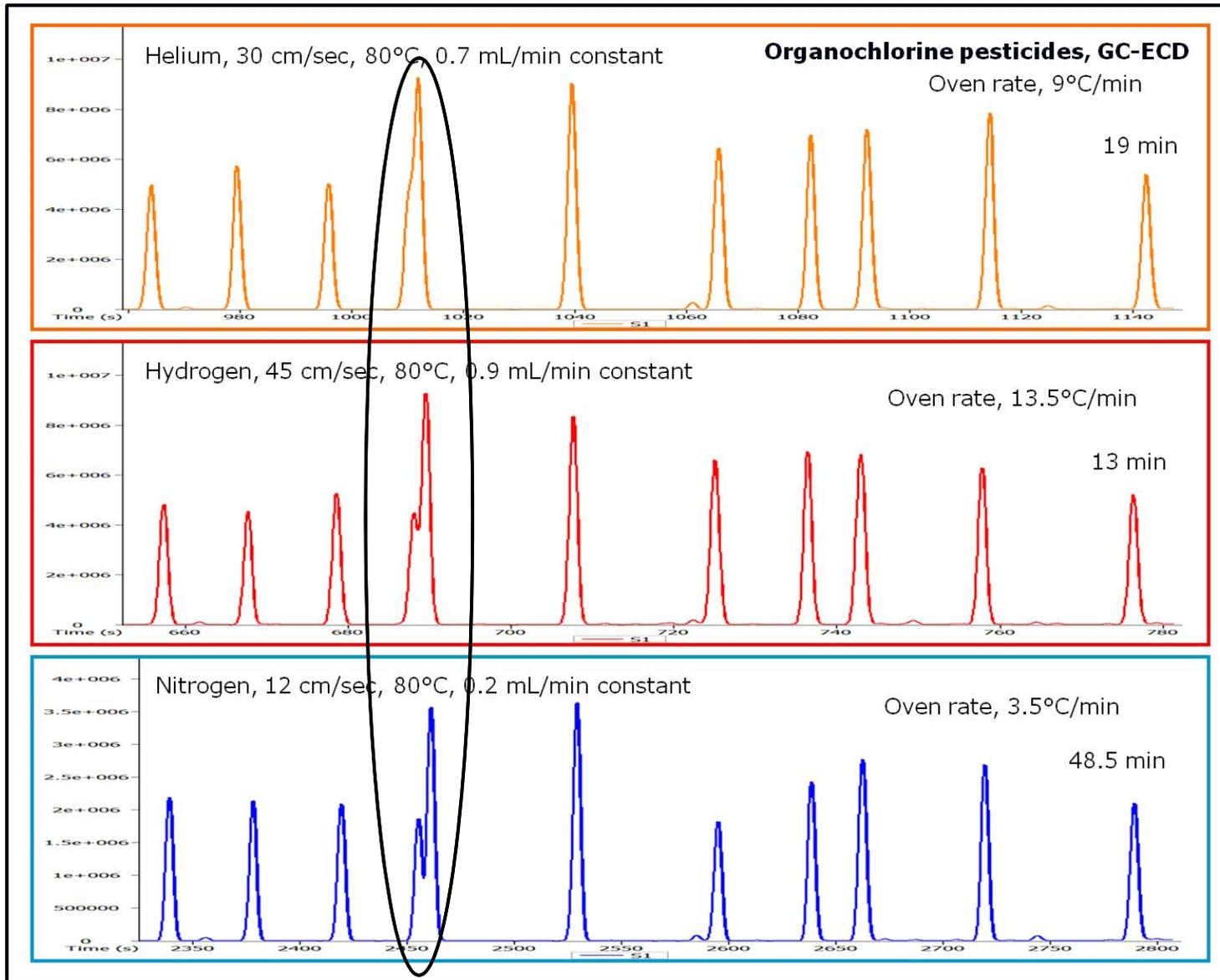
Van Deemter Plot



Efficiency Dependence on Column ID



OC Pesticides with different carrier gasses



3) Restek

Plan of Action

- **Goal: Minimize nitrogen in the source**
 - Helps reduce loss of sensitivity
 - Minimize linear velocity
 - Improves the position on the efficiency curve
- **Minimize nitrogen presence in the manifold**
 - Narrow bore columns
 - Lower linear velocities

Columns

- **Narrow bore columns**
 - 0.18 mm ID Column
 - 20M, 0.18 mm ID Rxi-5ms, 0.18 μ M film
 - 0.15 mm ID column
 - 20M, 0.15 mm ID SH- Rxi-5ms, 0.15 μ M film
- **Low flow through IP causes band broadening**
 - Liner volume is ~870 μ L
 - At 0.3 mL/Min that is an issue
 - Did not try low volume liner
 - Concerns about vaporization volume / flashover

Flow through the Injection Port

- **High pressure injection**
 - Works, but has disadvantages
 - Time
 - Leaks
- **High split ratio / larger injection volume**
 - On a 20M, 0.15 mm ID column flow is 0.33 mL/min @60°C and 27 cm/sec
 - Used large volume injection
 - Seems to work the best
 - Can be used in combination with high pressure if necessary

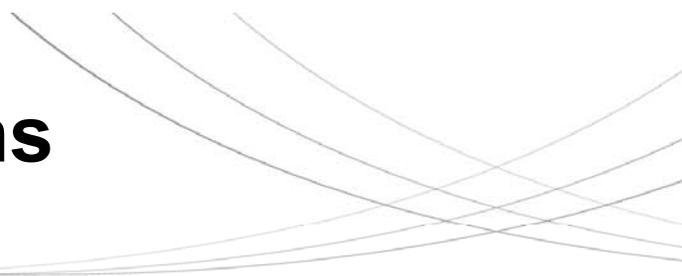
Surprises

- **Chromatography best at higher than typical linear velocity**
 - Most labs run helium at 35-45 cm/sec
 - Helium - 60 cm/sec on 0.18 mm ID Columns
 - Nitrogen - 45 cm/sec on 0.18 mm ID Columns
- **Temperature program is critical to PNA peak shape**
 - Run too high and you lose separation on the isobars – we all know that
 - Programming too fast is also a problem – we know that too
 - These effects are MUCH more pronounced with nitrogen – did not expect that

Work with 0.18 mm ID column

- **Bottom line: Did not work well.**
 - PNA peak shapes were problematic at low linear velocity
 - Was not able to compensate with temperature
 - Tried various injection techniques and liner types
- **Tried higher linear velocity**
 - Peak shape improved
 - Separation on the last few PNAs was not good enough
 - Signal was significantly attenuated on the late eluting PNAs
 - Signal was attenuated on the rest of the compounds too
 - Manifold pressure was well within specs $\sim 6.5 \times 10^{-6}$ Torr, but still high compared to helium (typically $\sim 8 \times 10^{-7}$ Torr @ 1 mL/min) on that instrument

MS Conditions for 0.18 mm Columns with Helium Carrier



GCMS-QP Series

Ion Source Temp. : °C
 Interface Temp. : °C
 Solvent Cut Time : min Detector Voltage : Relative to the Tuning Result Absolute
 Micro Scan Width : u kV
 Use MS Program : Threshold :

Group#1 - Event#1 GC Program Time : 22.75 min

	Start Time (min)	End Time (min)	Acq. Mode	Event Time(sec)	Scan Speed	Start m/z	End m/z	Ch1 m/z	Ch2 m/z	C
1	1.50	22.75	Scan	0.15	3333	35.00	500.00			
2	0.00	0.00	Scan	0.00	0	0.00	0.00			

GC Conditions for 0.18 mm Columns with Helium Carrier

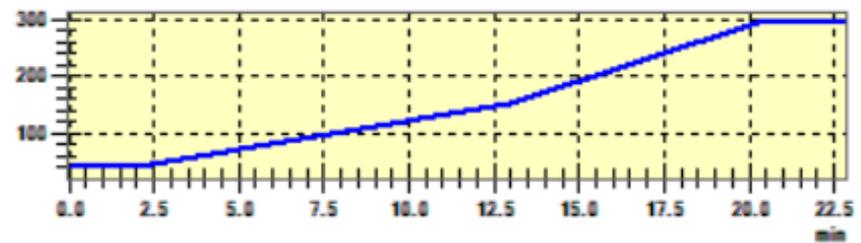
Inj. Port : SPL1 Inj. Heat Port : INJ1

Column Oven Temp. : °C

Injection Temp. : °C

Injection Mode :

Sampling Time : min



Carrier Gas : He Prim. Press. : 500-900

Flow Control Mode :

Pressure : kPa

Total Flow : mL/min

Column Flow : mL/min

Linear Velocity : cm/sec

Purge Flow : mL/min

Split Ratio :

Program :

	Rate	Final Temperature	Hold Time
0	-	40.0	2.00
1	10.00	150.0	0.00
2	20.00	295.0	2.50
3	0.00	0.0	0.00

Total Program Time : 22.75 min

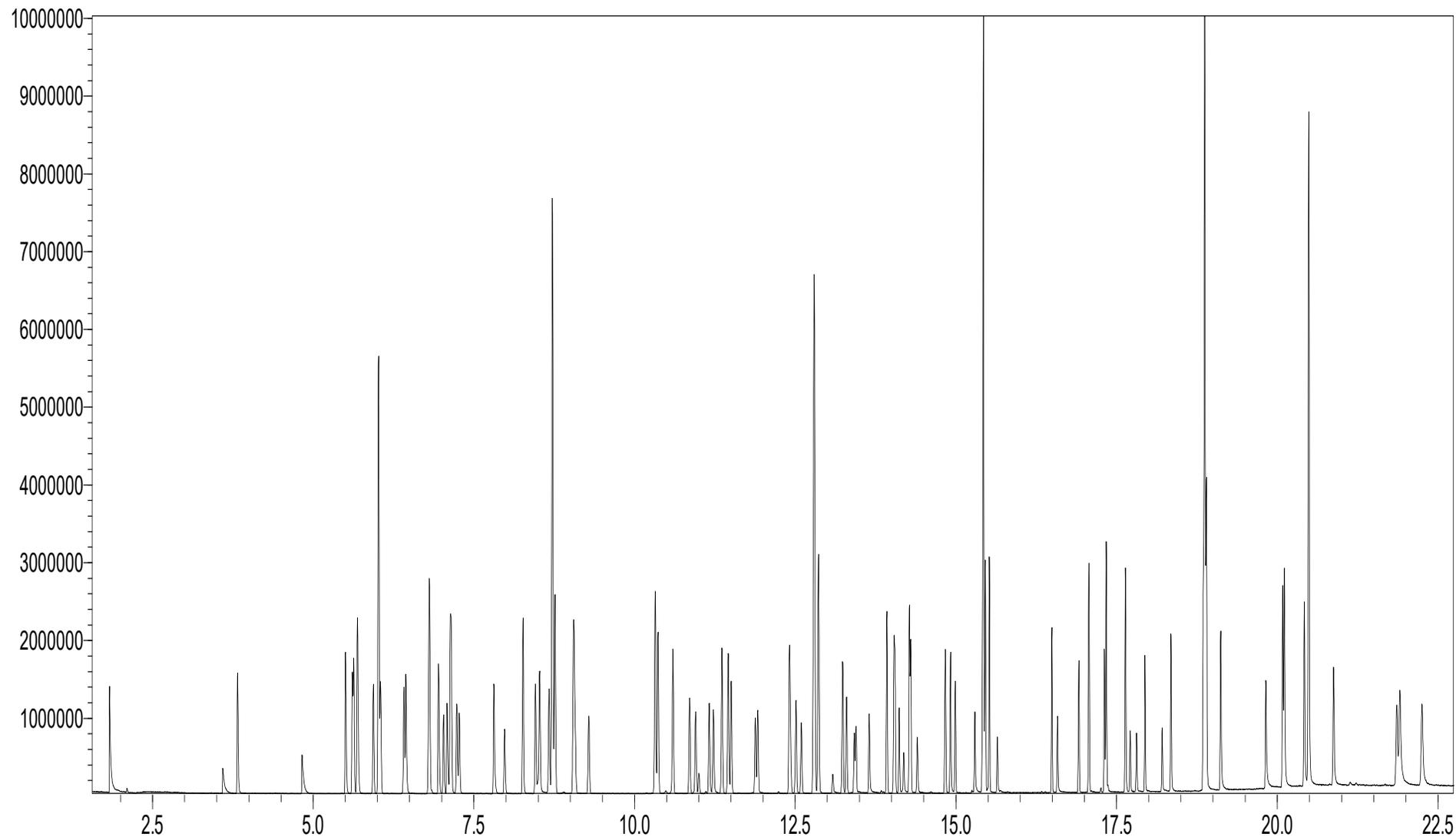
Column

Name Rxi5MS Thickness : 0.18 um

Length : 20.0 m Diameter : 0.18 mm

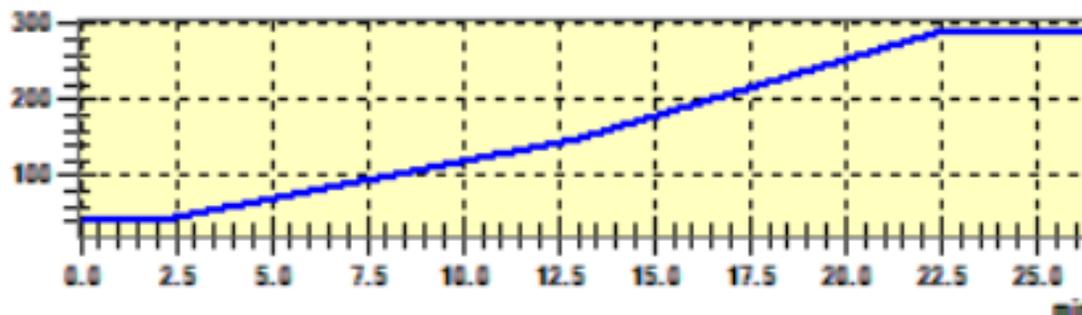
High Press. Injection Carrier Gas Saver

Chromatogram of a Standard with He Carrier



GC Conditions for 0.18 mm Columns with N2 Carrier

Column Oven Temp. : °C °C
 Injection Temp. : °C
 Injection Mode :
 Sampling Time : min



Carrier Gas : N2/Air Prim. Press. : 500-900

Flow Control Mode :

Pressure : kPa

Total Flow : mL/min

Column Flow : mL/min

Linear Velocity : cm/sec

Purge Flow : mL/min

Split Ratio :

Program :

	Rate	Final Temperature	Hold Time
0	-	40.0	2.00
1	10.00	150.0	0.00
2	15.00	290.0	4.00
3	0.00	0.0	0.00

Total Program Time : 26.33 min

Column

Name Rxi5MS Thickness : 0.18 um
 Length : 20.0 m Diameter : 0.18 mm

MS Conditions for 0.18 mm Columns with N2 Carrier

GCMS-QP Series

Ion Source Temp. : °C

Interface Temp. : °C

Solvent Cut Time : min

Detector Voltage : Relative to the Tuning Result Absolute

Micro Scan Width : u

kV

Use MS Program :

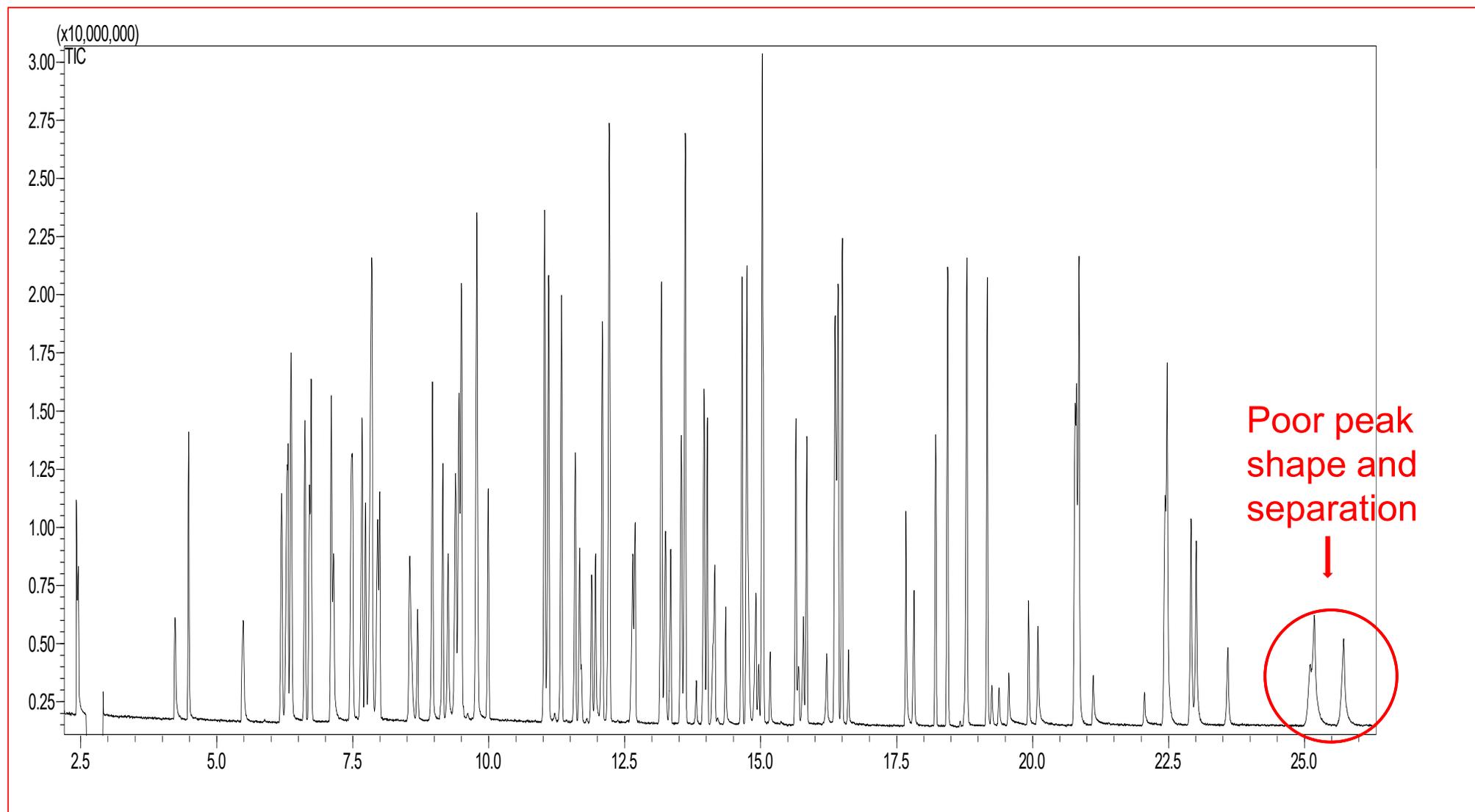
Threshold :

Group#1 - Event#1

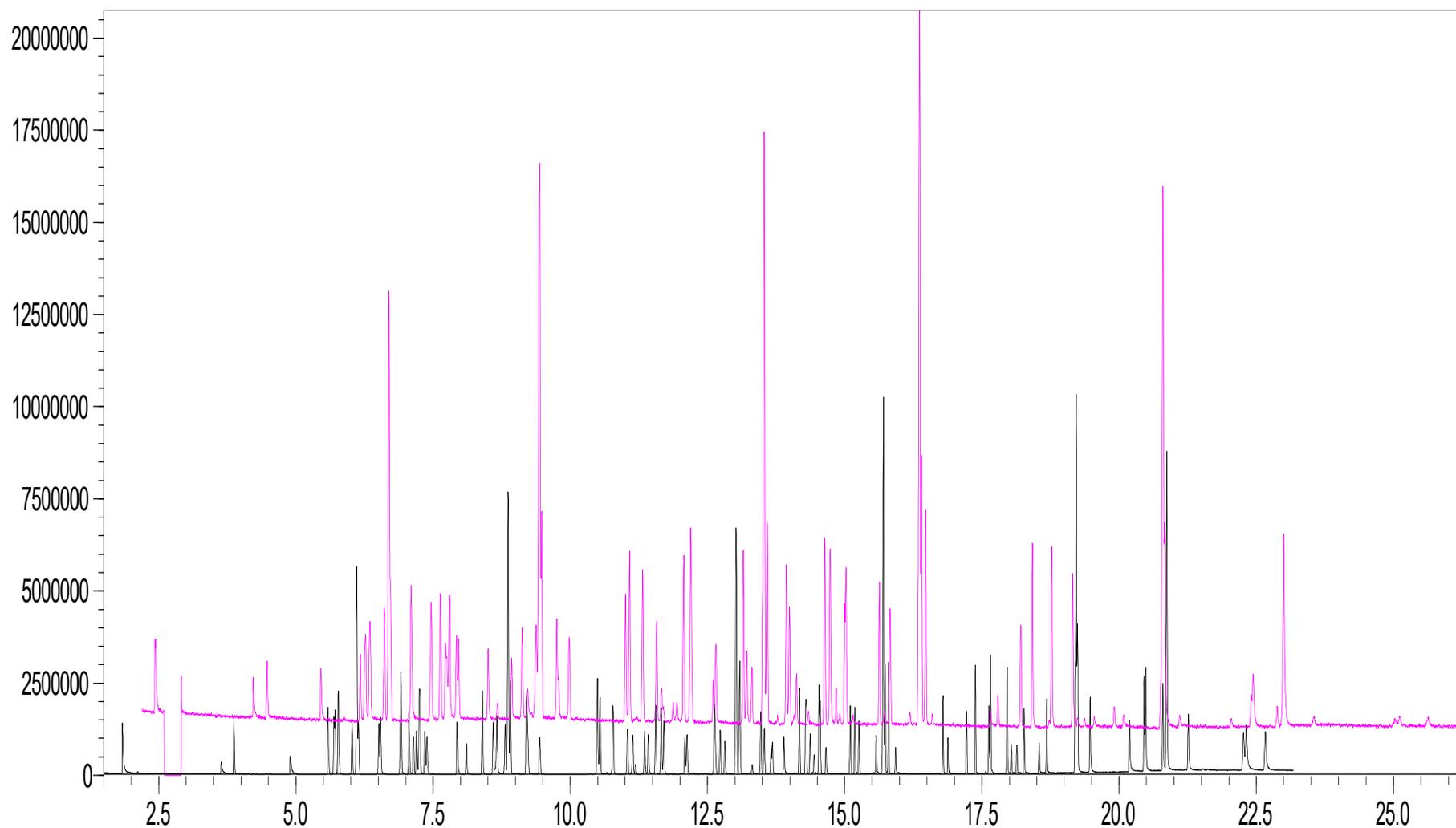
GC Program Time : 26.33 min

	Start Time (min)	End Time (min)	Acq. Mode	Event Time(sec)	Scan Speed	Start m/z	End m/z	Ch1 m/z	Ch2 m/z	
1	2.20	26.33	Scan	0.15	3333	35.00	500.00			
2	0.00	0.00	Scan	0.00	0	0.00	0.00			

Chromatogram of a Standard with N₂ Carrier on a 0.18 mm ID column



Overlay Chromatogram 0.18 mm ID He Carrier (black) and N₂ Carrier (pink)

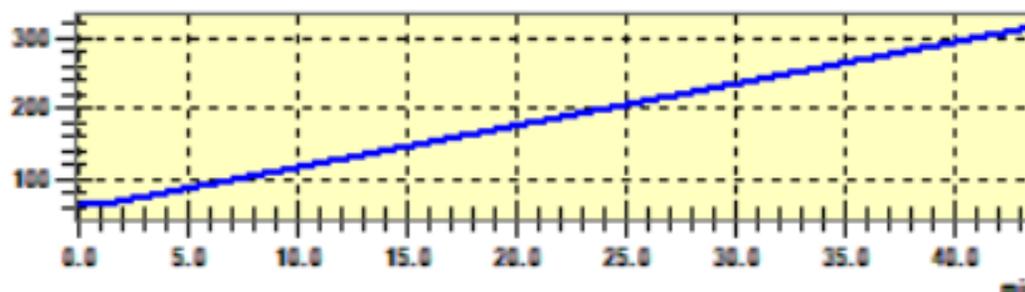


Work with 0.15 mm ID column

- **Chromatography was easier to control**
 - Linear velocity of 27 cm/sec worked well
 - Manifold pressure $\sim 1.5 \times 10^{-6}$ Torr
 - PNA peak shapes looked better
 - Was able to compensate for flow with temperature
 - Tried various injection techniques and liner types
 - Large volume split injections with a **Restek SkyBlue split liner** worked well
 - Finally settled on a 4 μ L injection with a 10:1 split

GC Conditions 0.15 mm ID Column

Column Oven Temp. : °C °C
 Injection Temp. : °C
 Injection Mode : ▾
 Sampling Time : min



Carrier Gas : N2/Air Prim. Press. : 500-900

Flow Control Mode : ▾

Pressure : kPa

Total Flow : mL/min

Column Flow : mL/min

Linear Velocity : cm/sec

Purge Flow : mL/min

Split Ratio :

Program : ▾

	Rate	Final Temperature	Hold Time
0	-	62.0	1.00
1	6.00	318.0	0.00
2	0.00	0.0	0.00
3	0.00	0.0	0.00

Total Program Time : 43.67 min

Column

Name SHRxi-5Sil MS Thickness : 0.15 um

Length : 20.0 m Diameter : 0.15 mm

MS Conditions 0.15 mm ID Column

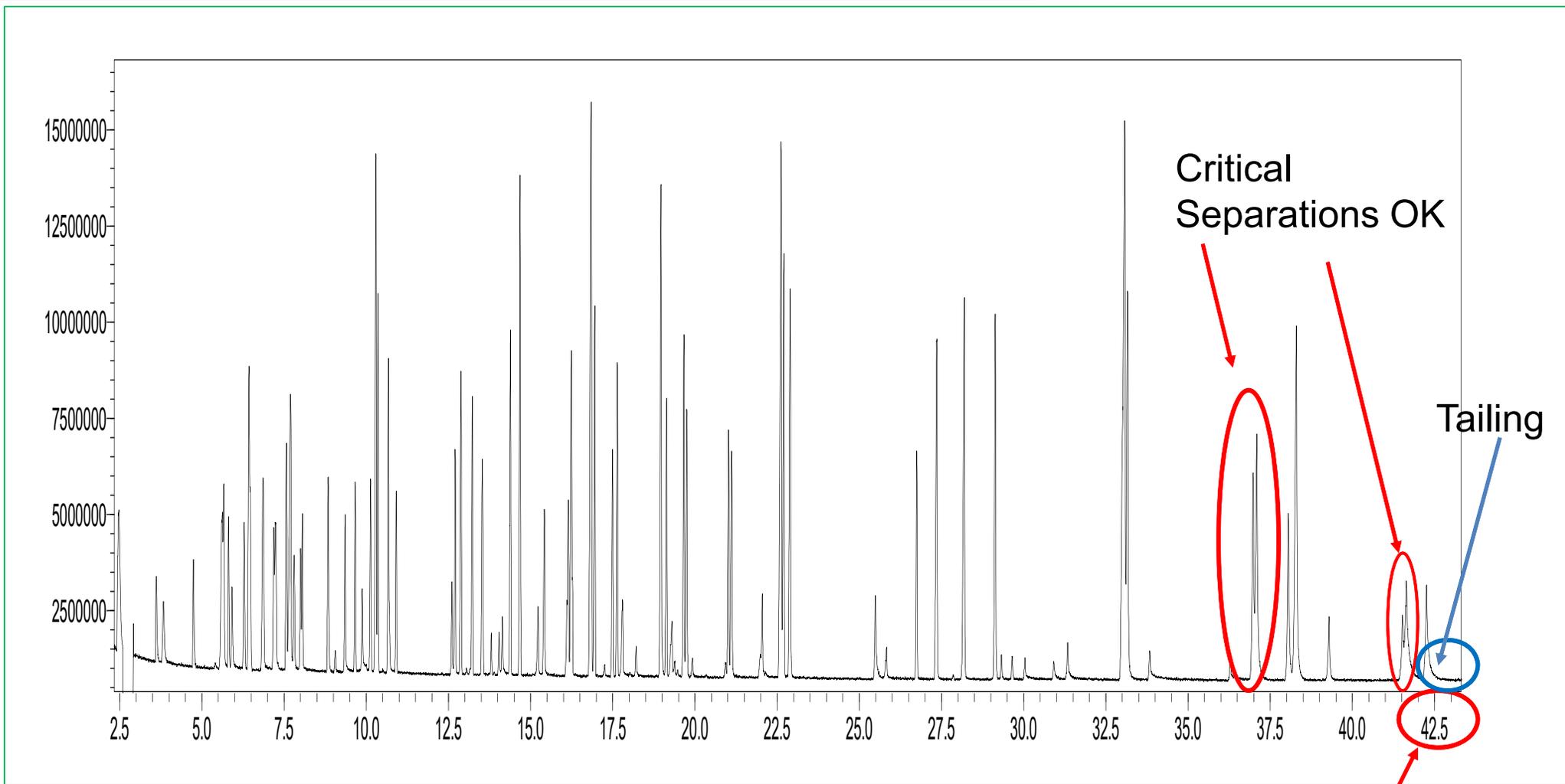
Ion Source Temp. : °C
 Interface Temp. : °C
 Solvent Cut Time : min Detector Voltage : Relative to the Tuning Resu
 Micro Scan Width : u kV
 Use MS Program : Threshold :

Group#1 - Event#1

GC Program Time : 43.67 min

	Start Time (min)	End Time (min)	Acq. Mode	Event Time(sec)	Scan Speed	Start m/z	End m/z
1	2.30	43.66	Scan	0.15	3333	35.00	500.00
2	0.00	0.00	Scan	0.00	0	0.00	0.00

Chromatogram 0.15 mm ID with N₂ Carrier



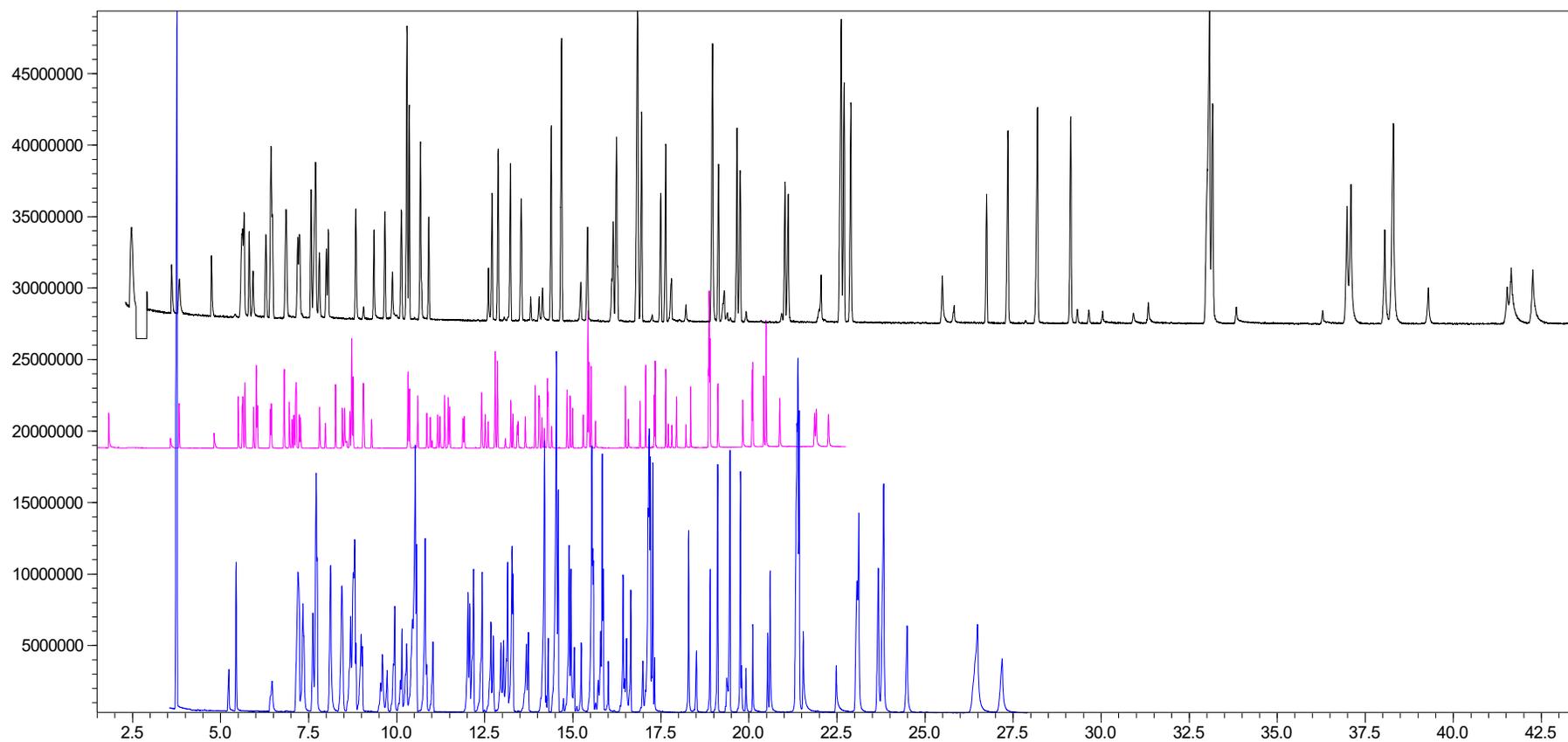
Note longer run time

Chromatograms

Black: Nitrogen on a 0.15 mm column

Pink: Helium on a 0.18 mm column

Blue: Nitrogen on a 0.18 mm column



Average IDLs on the 3 columns

Helium on the 0.18 mm ID column	0.096ng/ μ L
Nitrogen on the 0.18 mm ID column	0.555ng/ μ L
Nitrogen on the 0.15 mm ID column	0.331ng/ μ L

Other Issues We Encountered

● **Linearity**

- We were trying to run from 0.5 ng/μl to 100 ng/μl. Across that range many of the calibration curves were quadratic. That may not be the case over a smaller range (e.g. 20ng to 160ng)

● **Minimum RF**

- We failed to meet minimum RF criteria on half of the compounds that had defined limits. That may be a result of attempting to minimize detection limit.

● **Tuning**

- Factory tuning algorithms for tuning with N₂ carrier are not as well developed as for helium. Manual intervention helped.
- Were able to meet Method 8270D requirements with a little manual intervention, but not the older Method 625 criteria

Tuning – Method 8270D criteria

DFTPP KEY IONS AND ION ABUNDANCE CRITERIA

Mass Ion Abundance Criteria

68	<2% of <i>m/z</i> 69
69	Present
70	<2% of <i>m/z</i> 69
197	<2% of <i>m/z</i> 198
198	Base peak or present
199	5-9% of <i>m/z</i> 198
365	>1% of Base Peak
441	<150% of <i>m/z</i> 443
442	Base peak or present
443	15-24% of <i>m/z</i> 442

We were able to meet these criteria with little trouble. On some days, a little manual intervention was necessary.

Conclusions

1. **It seems that use of Nitrogen carrier is a viable option for the solid waste methods.**
2. **N₂ certainly works better than hydrogen for the “difficult” compounds**
3. **Detection limits and other results are encouraging**
4. **BUT... N₂ carrier unlikely to produce results equal to helium carrier**
5. **More work needs to be done**
6. **A work in progress**

Future Work

- **Linearity**
 - Even though the trend is toward lower detection limits, often detection limits are not an issue in the solid waste community.
 - Calibration curves will be analyzed over the narrow range typical of some solid waste labs.
- **Minimum RF**
 - A more concerning issue, but still one that may be alleviated by running higher level standards
 - Try columns with a higher phase ratio.
 - Also, try columns from different manufacturers
 - Try different liner geometries

Future Work

- **Run time**
 - It would be advantageous to work on shortening the run time, but not at the cost of decreased sensitivity
- **Tuning**
 - This is probably a matter of practice in the short term
 - Waiting for the engineers to improve the tuning algorithms in the long term.
- **Robustness**
 - Trying the system with actual extracts
 - Will it work in a “real world” lab?

Future Conversations

- **Has Method 8270 as we know it, reached an end point?**
- **Should it be broken up?**
 - “Easy” compounds to H₂ carrier.
 - Others to different techniques
- **Should we go to GC/MS/MS for some parts?**
 - MRM databases are ready now.
- **Should we go to LC/MS/MS for other parts?**
 - The “difficult” GC compounds are mostly acids and bases
 - Many (most?) should light-up with ESI or ACPI

References

1) **“Evaluation of Hydrogen as a Carrier Gas for Gas Chromatography / Mass Spectrometry”**

Shimadzu Application News No. SSI-GCMS-1303 February 2013

2) **EPA Method 8270D Analysis Using Narrow-bore GC Columns and Fast Data Acquisition with a Quadrupole GCMS System**

Richard Whitney, Ph.D. GC/GCMS Senior Product Specialist; Zhuangzhi “Max” Wang, Ph.D. GC/GCMS Product Specialist; Clifford M. Taylor, GC/GCMS Product Manager; Shimadzu Scientific Instruments, Columbia, MD, USA

3) **Nitrogen Carrier Gas for GC – Is it Feasible? – Is it Practical?**

Restek Chromatography, Jack Cochran, 2012

4) **Nitrogen as a Carrier Gas for Capillary GC**

LC/GC’s CHROMacademy

References

5) “Is Hydrogen the Best Carrier Gas for GC?”

5) LC/GC’s CHROMacademy, Dawn Watson

6) SW-846 Test Method 8270D: Semivolatile Organic Compounds by Gas Chromatography/Mass Spectrometry

7) Changing from Helium to Nitrogen and Maintaining the Separation Efficiency in the Same Analysis Time

Jaap de Zeeuw¹ and Jack Cochran,² Restek Corporation, Middelburg, The Netherlands and ²Restek Corporation, Bellefonte, US



Questions?

